



# The Future and Present of Thermal Analysis of Aluminum

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#### Outline

- Introduction
- Thermal analysis principles
- Data analysis with derivatives and integration
- Aluminum approach and examples
- Ductile iron approach and examples
- Conclusion
- Questions



# Who are you?

- Little poll to adjust my presentation
- What type of foundry are you?
  - Ductile and grey iron ?
  - Aluminum (low pressure) ?
  - Aluminum (high pressure) ?
  - Steel, allied steels, wear resistant iron ?
  - Magnesium ?



# The company

- Foundry Solutions & Metallurgical Services Inc.
  - We're metallurgists working with foundries and casting users
  - Foundry Services and consultants
    - Process Improvement
    - Project leader
    - Training
  - Distributor and Services provider of

Thermal Analysis with



Design and Process simulation Software with





#### AFS Conference theme

- Foundry challenges and futures in western countries
  - More competition coming from China and India
  - Decrease its cost and increase casting quality
  - Decrease delay and offers good customer services
  - Workers qualified less available
  - Investment in new equipment and process improvement
- What makes a Foundry in the communication age ?
  - Open-minded for new ways to do and adapt yourself
  - The world is your market, and the world is already here.
  - Real-time follow up your casting, quality, delivery and give excellent service to your client
  - Be Flexible and have fast delivery when needed
  - Use fast prototype and 3D technology



# New basics for a modern foundry

- What should do a modern foundry?
  - Should use technology to measure and control every process variables
    - Wireless technology, connected Device, On-Line probes
    - Modern analysis equipment (As Thermal analysis system)
    - Updated parameters for operator
    - Process control (SPC)
  - Should collect and analyze data
    - Database and quality indicators
    - Big data analysis
  - Should have an efficient quality system
    - Casting defect and root cause analysis
    - Training team with new method and parameters



# Thermal Analysis- a quality tool

- It's a powerful tool to measure and predict melt quality and casting defects.
- It's the fingerprint of the solidification process, and best allied of metallurgists.
- It can be used as quality control to adjust melt before pouring
  - Chill measurement, nucleate potential, shrinkage tendency (iron)
  - Modification and grain refinement levels (aluminum)
- It can be used as a process tool to adjust addition or find the best product. There is **potential saving** in your melt.
  - Additives as refiners, nodularizers, modifiers, inoculants
  - Adjustment of foundry return or second grade alloy





# Thermal Analysis- Principles

- As the metal turns solid, all the heat energy that went into melting is released.
- Every transformation has different energy release (or absorb) that affect cooling curves.
  - Number of seeds, grain sizes, dendrites formation
  - Phase transformation
    - Liquidus, eutectic, solidus
    - Perlites, carbides, graphite forms
    - Beta crystal, Al-Si, MgSi, AlCu, etc.
  - Exothermic and endothermic events (shrinkage, oxides)
  - Even small event like void, gas or bubbles
- It works for any metal!



# Base Iron chemistry Analysis

- Oldest use of TA (since 1931)
- It needs a tellurium cup (Ask for new Meltcup®)
  - Tellurium will prone white iron solidification
- Carbon equivalent, carbon and silicon are measured
- Carbon by TA is more accurate than by spectrometry. About equivalent to combustion analysis.
- Silicon needs to be adjusted by Spectro but it's fastest. Can be used when Spectro is down.







#### Meltlab Aluminum TA Stand

- Stand design for foundry better than old Pechiney Stand
- Same reusable steel cup to hold molten aluminum
- Reusable thermocouple probe and easier to change
- Quartz glass tube for sleeves instead of steel (cheaper)
- More intuitive handle
- Very simple thermocouple holder

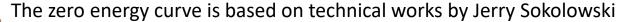
To ensure reliable results, we need a stand robust for the foundry.





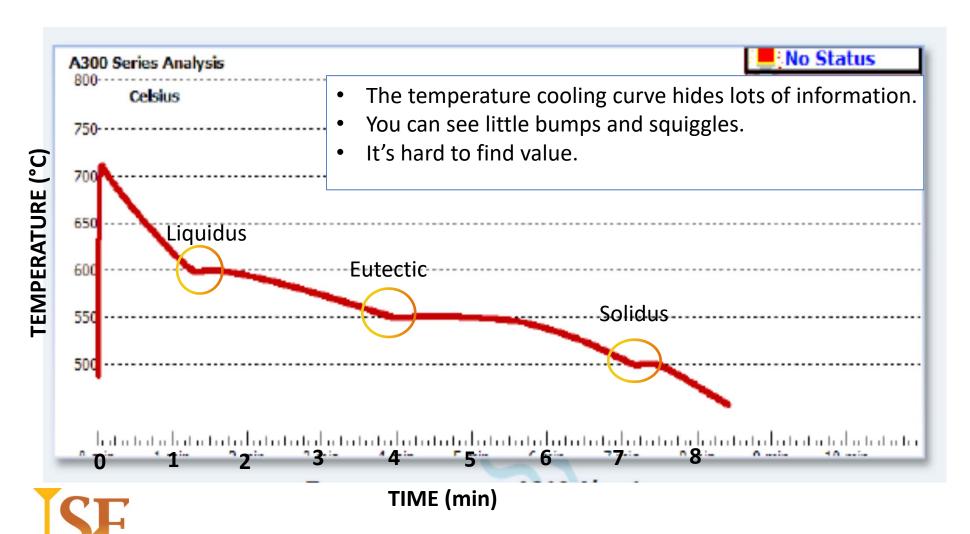
#### Derivatives are the microscope of TA

- Calculus, derivatives and integration, was invented by Newton.
- We use them to extract information from data in a reliable way automatically (by programming).
- Each derivative increase the magnification of variations in temperature showing smaller and smaller precipitants.
- The rate of cooling (1st derivative inverted) is used to find the strongest point in an arrest.
- Second derivative shows minor arrest and shrinkage activity
- Third, fourth and fifth derivatives are used to find the start and stop of a phase for integration purposes.
  - Heat proportion, phase proportion
- The zero energy curve is used with integration to calculation energy proportion of phases.

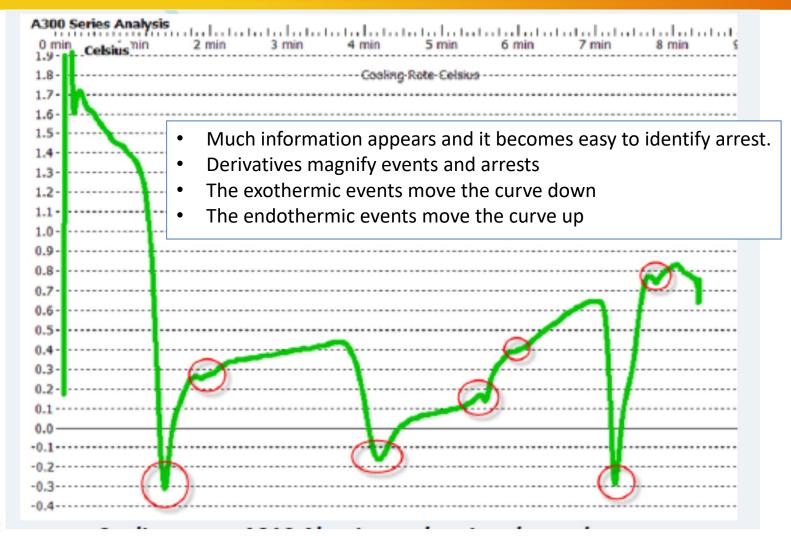


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# Temperature cooling Curve

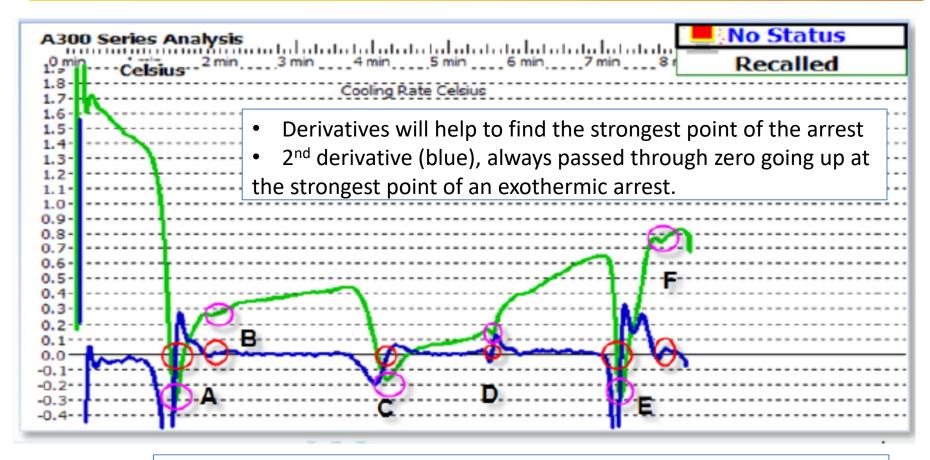


#### Cooling Rate Curve (1st derivative inverted)





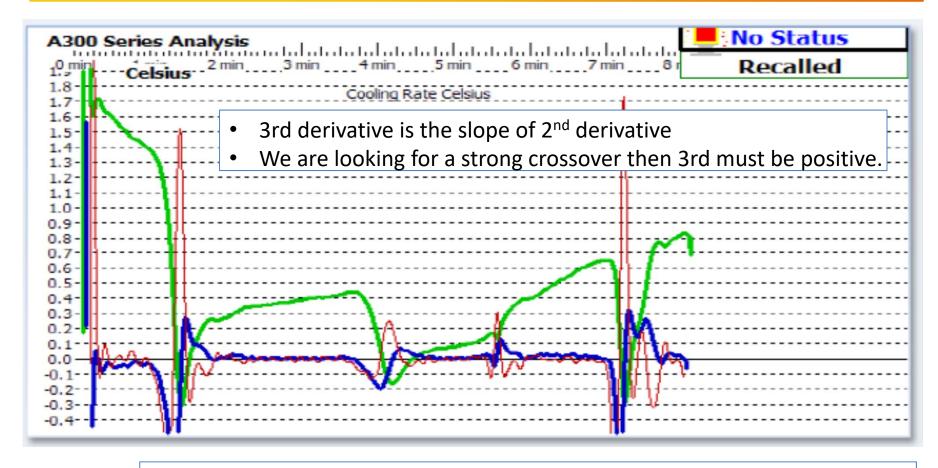
# Metrics by derivative

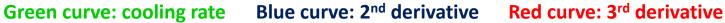


Green curve: cooling rate Blue curve: 2<sup>nd</sup> derivative



# Metrics by derivative

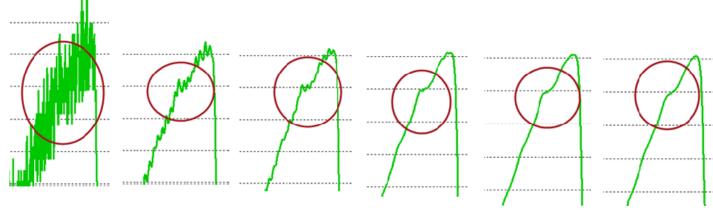






# Filtering and smoothing data

- How are we able to use derivatives?
  - Precision measurement with 16 resolutions of the T/C voltage.
  - Millivolts to temperature equations with 0.01% error.
  - Noise suppression in the electronics.
  - Unique data smoothing techniques (Biggest Meltlab secret)
  - Fourth and fifth derivatives for pinpointing the starting and ending of arrests.



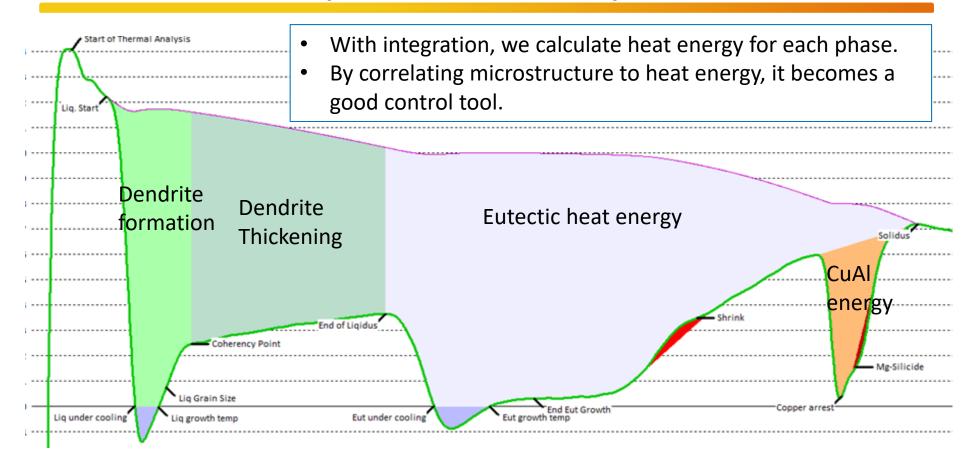


# Aluminum approach

- Control additions by measuring the degree of inoculation and degree of modification.
- Compensate for different charges mixes virgin and mixed returns with proper additions.
- Detect and measure shrinkage occurring with different procedures and different mixes.
- Detect gas bubbles forming in the metal sample and adjust process
  - Increase degassing or new alloy addition
- Control microstructures by using heat energy proportion for each phase.
- Keep all control parameters in range.



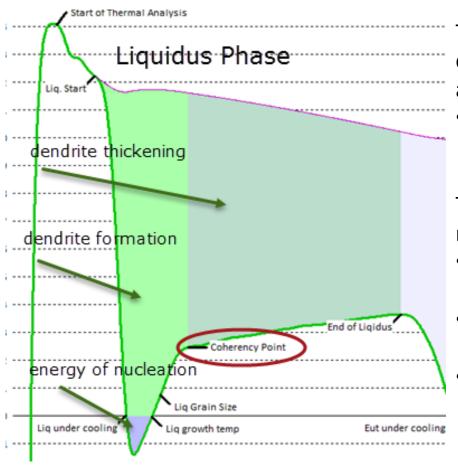
# Aluminum (A300 series)





Green Curve: Cooling Rate Magenta: zero energy curve

# Aluminum (A300 series)



The **coherency point** is where the dendrite crystals start interlocking with each other and begin to block feeding.

 Higher dendrite formation energy provides better feeding

This example is *under inoculated* and quickly reaches this interlocking form.

- The energy of nucleation is an inverse indication of inoculation.
- More inoculation would give better feeding characteristics for this metal.
- Use TA to control and stabilize your TiB additions.

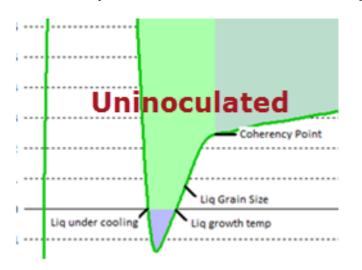


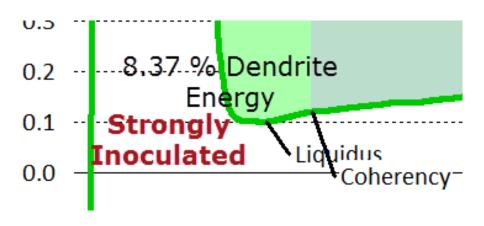
**Green Curve: Cooling Rate** 

Magenta: zero energy curve

### Aluminum-Inoculation (Grain refiners)

- Under inoculated aluminum produces large dendrites that decrease fluidity and blocks feeding.
- Liquidus has undercooling with less nucleation
- Fine dendrites become with inoculation
- Fluidity is better and coherency point is later.

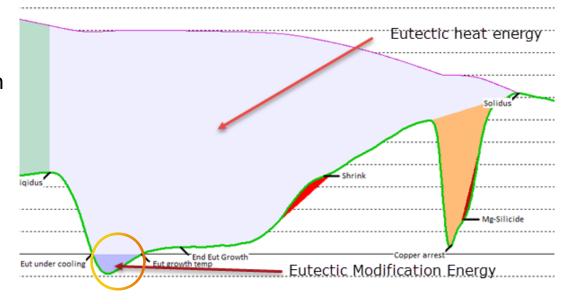






# Eutectic Phase Hypoeutectic A300

- The eutectic heat energy is a measure of the aluminum-silicon phase.
- As silicon levels increase, the percent of eutectic heat energy also increases. This is a control point.

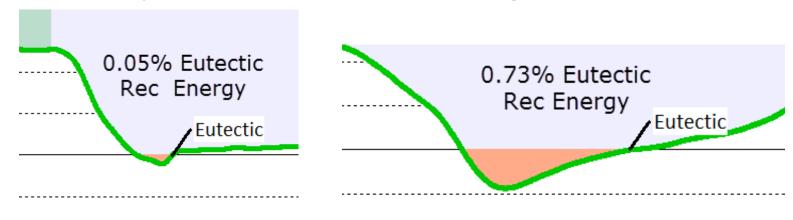


- The eutectic modification energy (blue area) is due to the degree of modification you have.
- Modification changes over time and with the different modifiers.
- Stabilize and control this with strontium or sodium additions to effect late feeding, decrease shrink and decrease heat treatment cost.



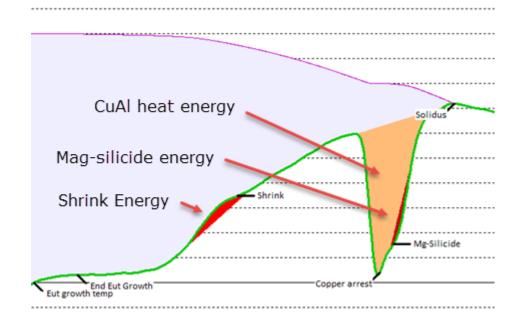
# Aluminum- modification (A300)

- Modification lowers the freezing point of the eutectic allowing longer feeding times.
- The degree of modification is measured by both the freezing temperature and the degree of undercooling of the eutectic.
- Recalesence is one way to look at it. The amount of energy is another.
- Strontium is expensive. You must add the right amount.





# Grain boundary phase (A300)



#### **Shrink Energy**

Incorrect modification levels and nucleation levels can influence the degree of shrinkage.

- The copper phase (CuAl) heat energy represents the copper that must be redissolved by heat treatment.
- This affects the strength you can achieve in the casting.
- The mag-silicide heat energy represents the amount of wearresistant material you have in your casting.
- Magnesium burns out easily and may need replenishing to maintain this arrest.



# Aluminum- future development

- SF would develop a new tool based on TA to predict porosities with criterias of hydrogen and oxides level in Aluminum.
  - As you've seen, our system is very sensitive.
- A professor of Metallurgical engineering from Laval University sponsor us for this project
- Trials we'll be done in university and technology center.
- Meltlab will introduce the new tool into Aluminum system
- We're waiting about project and subvention approval (or not) soon.

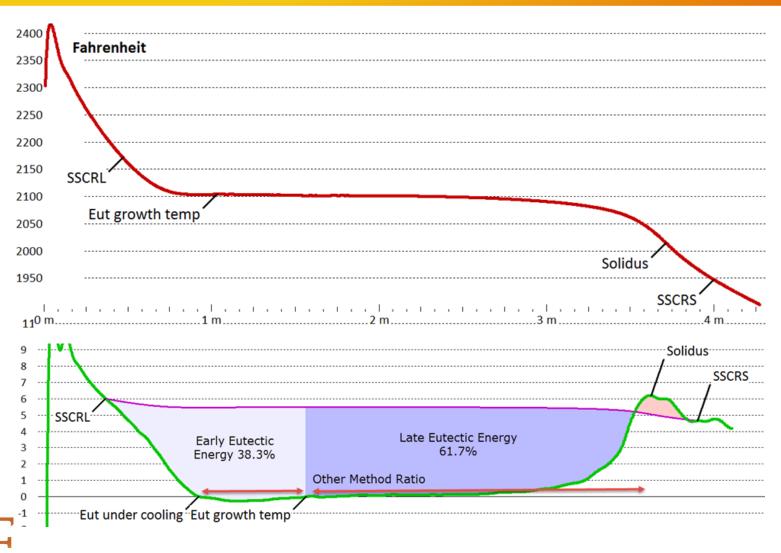


# Ductile iron approach

- Base iron chemistry: right amount of carbon and carbon equivalent.
- Base iron nucleation- nucleation potential change with time.
- Inoculation effectiveness the right amount improves nodular count and reduces shrinkage
- Magnesium Treatment the right amount improves nodular shape, reduces shrinkage, and prevents carbides. This needs balancing with inoculation.
- Nodularity when magnesium is ideal, and sulfur controlled.
- Nodule Count when inoculation is effective.
- Shrinkage minimized when everything is well balanced, iron is clean, and tramps controlled.



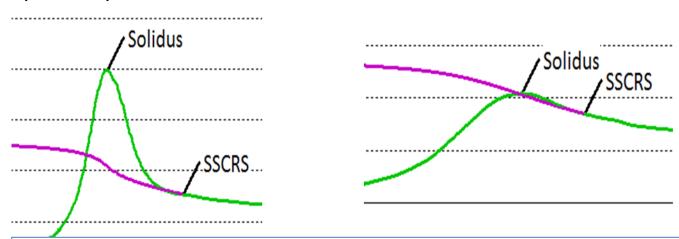
### Ductile iron basic curve



# Shrinkage in ductile iron

**Green Curve: Cooling Rate** 

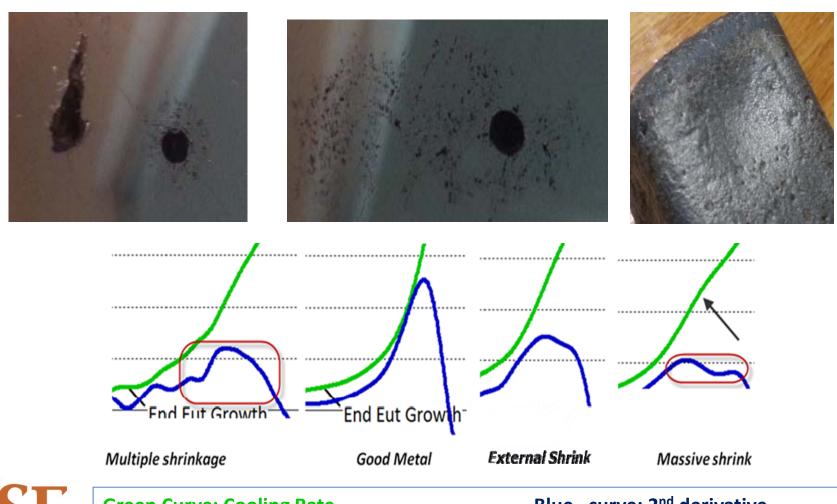
- Thermal analysis can detect the tendency of a sample of iron for micro and macro shrinkage, gas bubbles, and suck-in.
- We look for the presence of sufficient stress. If the stress has been reduced or totally removed by the time the casting has initially solidified, we know we have had a problem.
- The missing volume has been replaced with a shrink, gas holes or external shrink (suck-in).





Pink curve: zero energy curve

# Shrinkage in ductile iron





Green Curve: Cooling Rate Blue curve: 2<sup>nd</sup> derivative

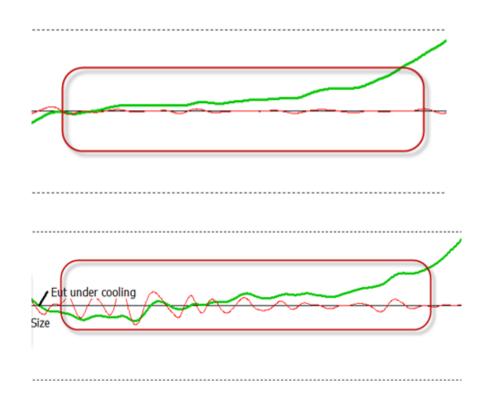
# Nodularity in ductile iron

Can you see any difference?

Top curve is 95% nodularity

Bottom curve is 85% nodularity.

Smoothness tests on the 3<sup>rd</sup> derivative can calculate the difference.

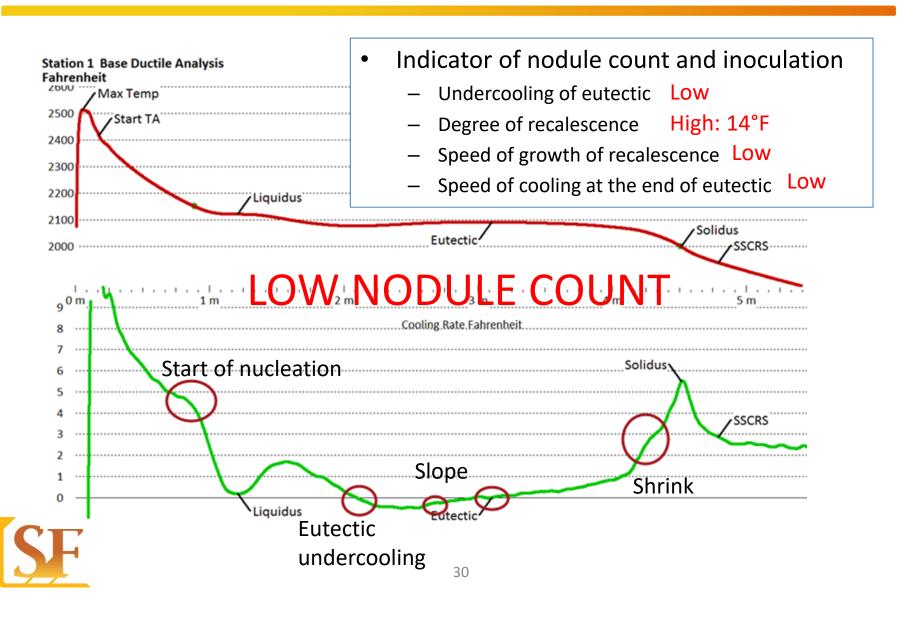




**Green Curve: Cooling Rate** 

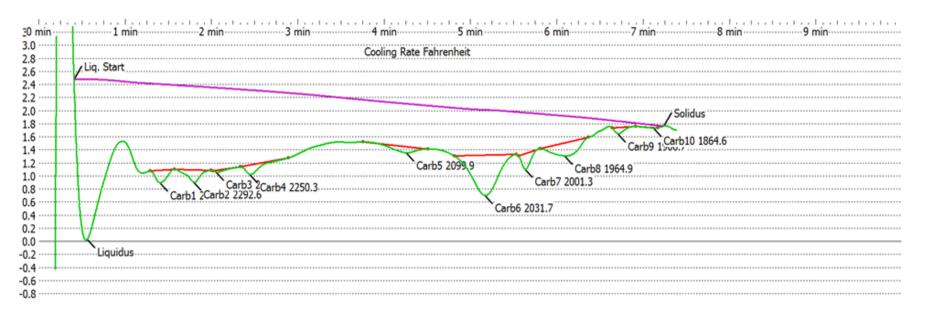
Red Curve: 3rd derivative

### Nodule count in ductile iron



#### Tool Steel- Before refinement

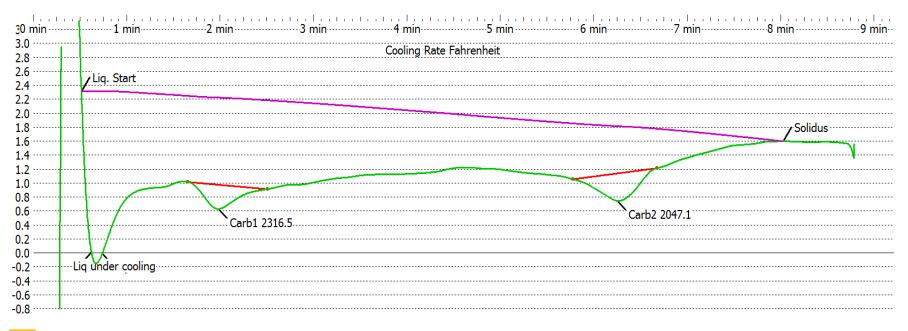
- Meltlab can detect many carbides in this allied steel
- Before refinement there are up to 10 different forms of carbides
- To know each composition, you would need deeper analysis with microstructure, electron microscope, XRF analysis or technical paper research.





#### Tool Steel- After refinement

- After refinement with both FeTi and FeV, there are only two forms of carbides
- You can have energy proportion of each carbides
  - A correlation with mass proportion or by microstructure is easy to program.





#### Conclusion

- There are differences in the metal that change with time and treatment.
- TA measures many critical variables for your melt quality.
- The most important is to find the "sweet spot" where metal has proper inoculation, effective solidification and good feeding to make good casting.
  - With best product and better cost for you
- Don't forget, without measures, you can't improve your process!
- To make consistent castings, start with a consistent metal.
- TA and Meltlab is the tool you must have for a modern foundry.



#### References

- Microstructure by Thermal Analysis AFS11-068, David Sparkman
- <u>A novel Algorithm for the calculation of Latent Heat of Solidification for Multi-component Aluminum Alloys</u>, AFS03-094, Jerry H. Sokolowski
- Thermal Analysis of Al casting alloy, AFS Cast Expo presentation, AFS16-105, Geoffrey K Sigworth
- The modification of Al-Si Casting Alloys- Important Practical and Theoretical Aspects- AFS08-019, Geoffrey K Sigworth
- <u>Improving Inoculation of Ductile Iron</u>- AFS03-103, S.Lekahk

You'll find plenty information about Thermal Analysis and Meltlab system by David Sparkman on <a href="https://www.meltlab.com">www.meltlab.com</a>





# Questions?



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